UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS

GCE Advanced Subsidiary Level and GCE Advanced Level

MARK SCHEME for the May/June 2012 question paper

for the guidance of teachers

9701 CHEMISTRY

9701/32 Paper 32 (Advanced Practical Skills 2), maximum raw mark 40

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

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International Examinations

Question	Sections	Indicative material	Mark
1 (a)	PDO layout	**I** Constructs a table for results (minimum of 3 results).	1
	PDO recording	**II** Appropriate headings and units for data given. Volume/V in cm^3 or / cm^3 or (cm^3) Time/t in seconds or / s or (s)	1
	PDO recording	**III** All times recorded to the nearest second.	1
	MMO decision	**IV** 3 additional volumes chosen with intervals not less than 2.00 cm^3 and all volumes of **FB 1** greater than or equal to 6.00 cm^3	1
	MMO collection	**V** In all 3 additional experiments water is added to make a total of 20.00 cm^3	1
	MMO quality	Round times to nearest second. **VI + VII** Compare time for 20.00 cm^3 of **FB 1** with that of supervisor.	2
		VIII + IX Compare time for 10.00 cm^3 of **FB 1** with that of supervisor. The range for award of 1 or 2 depends on the supervisor value.	2
		Supervisor value: < or = 15 δ for 2 is 2 and for 1 is 4 16 to 30 δ for 2 is 3 and for 1 is 6 31 to 45 δ for 2 is 4 and for 1 is 8 46 to 60 δ for 2 is 5 and for 1 is 10 > 60 δ for 2 is 6 and for 1 is 12	[9]
(b)	PDO display	**(i)** Working to show ans = 1×10^{-4} mol. Expression and answer.	1
	ACE interpretation	**(ii)** 0.5 × ans to **(i)** = 5×10^{-5} mol AND **(iii)** 2 x ans to **(ii)** = 1×10^{-4}	1
	PDO display	**(iv)** (1 x 10^{-4}) / 0.060 = 1.67×10^{-3} mol dm^{-3}.	1 [3]
(c)	ACE interpretation	Minimum of 4 results – rate correctly calculated in each case using ans **(b)(iv)** × 10^6 / time (or 2.25×10^{-3}). Min 2 s.f. rounded correctly.	1
	PDO recording	Units for rate given as mol $dm^{-3}s^{-1}$	1 [2]

Question	Sections	Indicative material	Mark
(d)	PDO layout	**I** Rate on *y*-axis and volume on *x*-axis. Axes clearly labelled	1
		II Linear scale chosen to use at least half of each axis (need not include 0,0) If no point at 0, 0 cannot count for > half.	1
		III Plotting of points. Minimum of 3 readings.	1
		IV Draws a line of best fit. Minimum of 4 readings including 0, 0 (if plotted).	1
(e)	ACE conclusion	Rate is (directly) proportional to Fe^{3+} **concentration.** Rate increases as concentration (volume) increases would score one	2 [2]
(f)	ACE	**(i)** 2 × 0.05 / 0.1	1
		0.10 / 20.00 × 100 = 0.5% 0.25 scores 1 mark. No ecf.	1
		(ii) Difficult to judge colour change / measurement of reaction time / some thiosulfate reacting with acid / formation of (S) ppt / variation in T.	1
	ACE improvement	**(iii)** Investigate reaction between Fe^{3+} and $S_2O_3^{2-}$	1 [4]
(g)	ACE conclusion	**(ii)** Thiosulfate concentration / number of moles / volume is halved (1) Time is shorter / reaction is faster with less thiosulfate (1) ora.	2 [2]
			[Total: 26]

Question	Sections	Indicative material			Mark
FB 5 = H_2SO_4; **FB 6** = K_2CrO_4; **FB 7** = $BaCl_2$; **FB 8** = $Pb(NO_3)_2$; **FB 9** = $NaNO_2$					
2 (a)	MMO collection	Orange **solution** (1)	White ppt **and** (white ppt in RH column) Yellow/cream ppt (1)	White ppt (1) Yellow ppt (1) White ppt (1)	[5]
Ignore excess of any reagent.					
(b)	ACE conclusion	Pb^{2+} in **FB 8** **AND** Ba^{2+} in **FB 7** H^+ in **FB 5** **AND** CrO_4^{2-} in **FB 6** SO_4^{2-} in **FB 5** **AND** Cl^- in **FB 7**			1 1 1 [3]
(c)	MMO decision	**I** **Warms** with NaOH and Al in **(i)**.			1
	MMO decision	**II** Adds named (dilute) acid in **(ii)**.			1
	PDO recording	**III** Presents observations in a single table – no extra reagents in **(iii)**.			1
	MMO collection	**IV** Ammonia / gas turns litmus blue in **(iii)**. If ammonia mentioned first, assume it is the gas that affects the litmus.			1
	MMO collection	**V** Brown fumes (of NO_2) / gas that turns blue litmus red in **(iii)**.			1
	ACE conclusion	**VI** nitrite (needs evidence).			1 [6]
					[Total: 14]